## (19) World Intellectual Property Organization

International Bureau



## 

(43) International Publication Date 6 January 2005 (06.01.2005)

**PCT** 

# (10) International Publication Number WO 2005/000273 A1

(51) International Patent Classification7:

A61K 9/14

(21) International Application Number:

PCT/IT2003/000401

(22) International Filing Date: 27 June 2003

27 June 2003 (27.06.2003)

(25) Filing Language:

Italian

(26) Publication Language:

English

- (71) Applicant (for all designated States except US): BIO-PROGRESS S.P.A. [YT/IT]; Via Aurelia 58, I-00165 Roma (IT).
- (72) Inventors; and
- (75) Inventors/Applicants (for US only): OLIVIERI, Aldo [IT/IT]; Via Aurelia 58, I-00165 Roma (IT). BO-NANOMI, Michele [IT/IT]; Via Brescia 29, I-00198 Roma (IT). PAZZI, Piergiorgio [IT/IT]; Via A. De Gasperi 3, I-66050 San Salvo (IT).
- (74) Agent: TANSINI, Elio, Fabrizio; Bugnion S.p.A., Viale Lancetti, 17, I-20158 Milano (IT).

- (81) Designated States (national): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW.
- (84) Designated States (regional): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE, SI, SK, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

#### Published:

with international search report

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(54) Title: COMPOSITE PRODUCT OBTAINABLE BY COGRINDING OF A ACTIVE PRINCIPLE WITH A COPOLYMER N-VINYL-2-PYRROLIDONE/VINYL-ACETATE

		Nimesulide /			Activation	time (hou	rs)	
Carrier	Example	Carrie <del>r</del> Ratio	0		1		2	
			ΔU <sub>t</sub> (mJ/mg)	T, (°C)	ΔH <sub>f</sub> (mJ/mg)	T <sub>f</sub> (°C)	ΔH <sub>1</sub> (mJ/mg)	T <sub>r</sub> (°C)
NVP/VA	3	1/3	60.6	137.3	34.7	114.9	26.0	108.5
NVF/VA	4	1/4	59.6	140.2	19.0	109.4	10,6	107.7
PVP	E	1/3	95.8	149.8	28.5	-136.0	21.0	135.4
PVP	F	1/4	75,9	149.4	15.0	133.0	15.1	130.5
PVP-CL	G	1/3	79.2	150.7	32.9	132.6	31.2	132.0
PVP-CL	н	1/4	77.2	150.5	24.6	130.6	23.7	129.6

(57) Abstract: The present invention describes how to obtain composite products comprising an active substance supported by a carrier, in which the carrier is the linear copolymer of N-vinyl-2-pyrrolidone with vinyl acetate. The composite products are obtained by co-grinding of the dry mixture of the active substance and of the aforesaid carrier. The composite products thus obtained have better chemical-physical properties (lower melting enthalpy and/or lower melting temperature of the active substance) and a higher dissolution speed with respect to composite products obtained with the same-co-grinding time with other carriers used in prior techniques. Furthermore, the composite products obtained with the technique according to the present invention have the appearance of powders that are easier to work from a pharmaceutical point of view (flow, compression) with respect to composite products previously obtained with other carriers.

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Composite product obtainable by cogrinding of a active principle with a copolymer N-vinyl-2-pyrrolido-ne/vinyl-acetate.

#### FIELD OF THE INVENTION

The present invention relates to a composite product obtained by co-grinding of an active principle with a carrier comprising a N-vinylpyrrolidone/vinyl acetate copolymer. The present invention further relates to the use of said composite product for preparing pharmaceutical compositions aiming at improving absorption properties (shorter times for appearance of haematic peak, higher bioavailability) of drugs or active substances scarcely soluble in environment.

#### 15 STATE OF THE ART

The improvement of absorption properties, i.e. of times for appearance of haematic peak and of bioavailability as the area under the haematic curve, of drugs or pharmaceutical active substances that are scarcely soluble in water environment, has been the object of several studies and quite different technical suggestions: micronization of the active substance with subsequent reduction of granulometry and increase of surface area; formulation with surfactants; complexation with cyclodextrins and derivates; co-precipitation or extrusion with linear polymers.

Among the most innovative techniques that can enable to obtain products with improved biopharmaceutical properties, co-grinding of the scarcely soluble active substance with hydrophilic carriers has been among the most applied solutions thanks to its interesting results.

The use of co-grinding of scarcely soluble drugs with

the water soluble linear polymer polyvinylpyrrolidone was described in 1975 (Chem. Pharm. Bull., 23, 2973, 1975). In later publications (Chem. Pharm. Bull., 78, 3340, 1977; Chem. Pharm. Bull., 28, 652, 1980) microcrystalline cellulose was used as carrier for cogrinding.

The use of beta-cyclodextrin was described in JP Patent 7986607 and in DE Patent 3427788, in which lactose, calcium phosphate and starch are mentioned as further materials for co-grinding, to be added to cyclodextrin, if necessary.

Hydrophilic silica gel and other adsorbing inorganic materials were described in EP 129893, in which the obtained co-ground products are characterized by amorphizations of active substances and improved dissolution properties.

US Patent 4,639,370 describes co-grinding of scarcely soluble drugs with reticulated polymers that are insoluble but can be swelled in water, such as reticulated polyvinylpyrrolidone or reticulated sodium carboxymethyl cellulose: evident improvements of the properties concerning passage into solution and of absorption properties are obtained for very scarcely water soluble drugs such as methyl hydroxy progesterone acetate.

WO 9632931 describes the use of starch sodium glycolate as carrier with improvement of dissolution speed and anticipation of time of appearance of ibuprofen haematic peak.

In all documents referred to above co-grinding is carried out, leaving aside the kind of grinding mill or carrier used, on dry mixtures of the active substance, the carrier being introduced into the grinding chamber of the selected device.

In US Patent 5,354,560 and US Patent 5,449,521 the mixture of powders to be co-ground are introduced into grinding chambers that have been pre-saturated with water vapors or vapors from solvents that can solubilize the active substance. Here lower co-grinding times for obtaining improved properties of scarcely soluble drugs are claimed.

The improvements made to co-grinding technique by the introduction of solvent vapors into the co-grinding chambers, however, are negatively counterbalanced by higher process costs and by greater problems involving residues in the final product and by higher possibilities of degradation of the active substance.

## DESCRIPTION OF THE INVENTION

It has now been surprisingly found that the use of linear copolymer N-vinyl-2-pyrrolidone/vinyl acetate as carrier for co-grinding of scarcely water soluble drugs results in improvements of the crystalline structure of said drugs (reduction of melting enthalpy and/or melting point), greater increase of solubility and of dissolution speed with respect to, co-grinding times being the same, what can be obtained by using other carriers commonly used for co-grinding, such as for instance linear polyvinylpyrrolidone, reticulated polyvinylpyrrolidone or cyclodextrins.

The technical features and the advantages deriving from the use of N-vinyl-2-pyrrolidone/vinyl acetate copolymer (hereinafter referred to as NVP/VA for reasons of shortness) as carrier for co-grinding of scarcely soluble drugs according to the present invention will be evident from the following detailed description.

In a preferred embodiment of the present invention the selected active substance and NVP/VA carrier, both in

powder form, are pre-mixed in a suitable powder mixer. Preferably, powder granulometry can vary within a range between 0.01 and 1,000 microns both for the carrier and for the drug; for instance it can vary

- between 0.1 and 200 microns.
  - Preferably, the mixture comprising the selected active substance and NVP/VA carrier is introduced into the grinding chamber of any grinding mill, together with the grinding means.
- Alternatively the mixture comprising the selected active substance and NVP/VA carrier can be introduced directly into the grinding mill without pre-mixing.
  - The mill that can be used for co-grinding an active substance with NVP/VA carrier comprises a grinding chamber housing grinding means of any kind (for instance balls or cylinders).
  - Co-grinding comprises for instance a mechanical stirring carried out by rolling, centrifugal rotation or vibration.
- 20 Co-grinding can be carried out at low or high energy for times varying from 0.1 to 48 hours; preferably for times between 0.5 and 8 hours.
  - Preferably, the weight ratio of NVP/VA carrier to active substance can be between 200:1 and 1:10; still more preferably between 100:1 and 1:5; for instance
- more preferably between 100:1 and 1:5; for instance between 10:1 and 0.5:1.
  - At the end of co-grinding the resulting powder comprising the composite product according to the present invention can be sieved or used directly in the preparation of the pharmaceutical composition in the desired pharmaceutical form, for instance tablet,
  - The pharmaceutical composition comprising the composite product is prepared by means of the

capsule, packet, powder, pellet, syrup or solution.

technique known to the person skilled in the art and by using excipients and pharmaceutically acceptable additives commonly used for preparing the desired pharmaceutical forms.

5 Many different classes of drugs can be usefully worked with the technique of the present invention, from anti-inflammatory agents to analgesics, relaxants, anti-microbic agents, antiseptics, acid inhibitors,  $\mbox{H}_{2}$  antagonists, anti-emetics and antinausea, biliary acids, oral hypoglycemizers, diuretics, anti-hypertensives, sulfonamides, inhibitors, hypolipemizers, anti-mycotic agents, antihistamines, hormones, quinolone derivates, antibacterial agents, beta-lactame and fluoroquinolone 15 antibiotics, antiviral agents, anti-neoplastic agents, immuno-modulators and immuno-suppressors, anti-gout agents, anesthetics, analgesics, antipyretics,  $5HT_1$ agents, anti-psychotic agonists, anti-Parkinson agents, tranquillizers, antidepressants, antiparasitic agents, non-cortisone anti-allergic agents, anti-asthmatic agents, anti-glaucoma inhibitors of carbonic anhydrase, beta-blockers, and others.

Drugs to which the present invention can be applied, whatever the therapeutic class they belong to, are scarcely water soluble drugs and drugs with low dissolution speed.

Non-exhaustive examples are paracetamol, nifedipine, diclofenac, sulindac, ibuprofen, piroxicam, 30 alclofenac, ketorolac, indomethacine, naproxen, fenoprofen, flurbiprofen, ketoprofen, cimetidine, ursodeoxycholic mesalazine, acid, ranitidine, mefenamic acid, sinvastatin, megestrol acetate, diazepam, cyclosporin, ubiquinone, lorazepam,

tolbutamide, ketanserine, furosemide, nicergoline, losartan, econazole, miconazole, taxol, progesterone, prednisolone, beclometasone, nalidixic acid, finasteride, ciprofloxacine, ofloxacine, lomefloxacine, methotrexate, etoposide, daunorubicine, tamoxifen, allopurinol, clodronic acid, sumatriptan, carbamazepine, clorpromazine, clozapine, sulpiride, buspirone, fluoxetine, citalopram, caffeine, metronidazole, acetazolamide etc.

10 N-2-vinyl-pyrrolidone/vinyl acetate copolymer is a flowing powder with a high pharmaceutical workability thanks to the spray-drying process used to obtain it. Said spray-drying technique results in spherical particles with a limited and homogenous size distribution. Said morphology of the particles positively affects the flow of the powder and its ability to mix with other excipients.

Another very important property of said copolymer is its low glass transition temperature (Tg).

20 By mere comparison, Tg of NVP/VA is of about 106°C, whereas the one of the corresponding linear polymer N-vinylpyrrolidone is of about 160°C.

Glass transition temperature can be defined as the temperature at which a polymer starts to get fluid without being completely melted.

The Applicant has found it useful to exploit the technical property of Tg of NVP/VA for preparing pharmaceutical forms for instance as tablets. As a matter of fact, during the compression of the polymeric powders a low glass transition temperature helps the formation of inter-particle bonds thanks to an easier deformation/fluidization of the single particles. The formation of inter-particle bonds enables to obtain a tablet pharmaceutical form having

a higher hardness than the one that could be obtained with other traditional excipients by compression, such as for instance lactose or microcrystalline cellulose. Said better compression properties of NVP/VA copolymer have led the Applicant to use successfully said copolymer as carrier for co-grinding of little soluble drugs so as to obtain powders of drug/carrier composite with improved compression properties with respect to commonly used carriers such as linear polymers, for instance polyvinylpyrrolidone, cyclodextrins, reticulated polymers, for instance crospovidone.

Beyond better compression properties NVP/VA copolymer has a lower hygroscopicity with respect to the corresponding linear polymer polyvinylpyrrolidone. This lower hygroscopicity involves a higher stability of the copolymer during its conservation (preservation of flow and compression properties) and a lower negative impact on the stability of moisture-degradable drugs mixed or better co-ground with said carrier.

Now, it has been unexpectedly found that beyond better compression properties of the resulting powders, coground products deriving from scarcely soluble drugs and NVP/VA have lower melting enthalpies and lower melting temperatures with respect to co-ground products obtained with previously used carriers such as polyvinylpyrrolidone, cyclodextrins, crospovidone. The following examples of co-ground products obtained with NVP/VA (examples 1-8) are given to a mere illustrative and non-exhaustive purpose for the invention, compared with co-ground products obtained with other commonly used carriers (examples A-O). The properties of co-ground products obtained are

described in tables 1-7.

### EXAMPLES

- 1 16.6 g of <u>nimesulide</u> are mixed with 49.8 g of <u>NVP/VA</u> for 15'. The powder is then poured into the grinding chamber of a low energy vibrational mill. Grinding is carried out for 3 hours.
- 2 13.3 g of <u>nimesulide</u> are mixed with 53.2 g of <u>NVP/VA</u> for 15'. The powder is then poured into the grinding chamber of a low energy vibrational mill.
  10 Grinding is carried out for 3 hours.
- 3 16.6 g of <u>nimesulide</u> are mixed with 49.8 g of <u>NVP/VA</u> for 15'. The powder is then poured into the grinding chamber of a low energy vibrational mill. Grinding is carried out for 2 hours.
- 15 4 13.3 g of <u>nimesulide</u> are mixed with 53.2 g of <u>NVP/VA</u> for 15'. The powder is then poured into the grinding chamber of a low energy vibrational mill. Grinding is carried out for 2 hours.
  - $\frac{5-}{2.5}$  g of <u>piroxicam</u> are mixed with 12.5 g of NVP/VA for 15'. The powder is then poured into the grinding chamber of a centrifugal mill. Grinding is carried out for 4 hours.
    - 6 180.0 g of <u>nifedipine</u> are mixed with 800.0 g of <u>NVP/VA</u> for 15'. The powder is then poured into the grinding chamber of a high energy vibrational mill. Grinding is carried out for 4 hours.
    - 7 13.3 g of ursodeoxycholic acid (<u>UDCA</u>) are mixed with 53.2 g of <u>NVP/VA</u> for 15'. The powder is then poured into the grinding chamber of a low energy vibrational mill. Grinding is carried out for 4 hours.
  - 8 11.1 g of ursodeoxycholic acid (<u>UDCA</u>) are mixed with 55.5 g of <u>NVP/VA</u> for 15'. The powder is then poured into the grinding chamber of a low energy vibrational mill. Grinding is carried out for 4 hours.

- A 16.6 g of <u>nimesulide</u> are mixed with 49.8 g of <u>PVP</u> for 15'. The powder is then poured into the grinding chamber of a low energy vibrational mill. Grinding is carried out for 3 hours.
- 5 <u>B 13.3 g of nimesulide</u> are mixed with 53.2 g of <u>PVP</u> for 15'. The powder is then poured into the grinding chamber of a low energy vibrational mill. Grinding is carried out for 3 hours.
- <u>C</u> 16.6 g of <u>nimesulide</u> are mixed with 49.8 g of <u>PVP-CL</u> for 15'. The powder is then poured into the grinding chamber of a low energy vibrational mill. Grinding is carried out for 3 hours.
  - $\underline{D}$  13.3 g of <u>nimesulide</u> are mixed with 53.2 g of  $\underline{PVP}$ -CL for 15'. The powder is then poured into the grinding chamber of a low energy vibrational mill. Grinding is carried out for 3 hours.
  - $\underline{E}$  16.6 g of <u>nimesulide</u> are mixed with 49.8 g of <u>PVP</u> for 15'. The powder is then poured into the grinding chamber of a low energy vibrational mill. Grinding is carried out for 2 hours.
  - F-13.3 g of <u>nimesulide</u> are mixed with 53.2 g of <u>PVP</u> for 15'. The powder is then poured into the grinding chamber of a low energy vibrational mill. Grinding is carried out for 2 hours.
- Q 16.6 g of <u>nimesulide</u> are mixed with 49.8 g of <u>PVP-CL</u> for 15'. The powder is then poured into the grinding chamber of a low energy vibrational mill. Grinding is carried out for 2 hours.
- H 13.3 g of <u>nimesulide</u> are mixed with 53.2 g of <u>PVP-CL</u> for 15'. The powder is then poured into the grinding chamber of a low energy vibrational mill. Grinding goes on for 2 hours.
  - $\underline{\text{I}}$  2.5 g of  $\underline{\text{piroxicam}}$  are mixed with 12.5 g of  $\underline{\text{PVP-}}$   $\underline{\text{CL}}$  for 15'. The powder is then poured into the

grinding chamber of a centrifugal mill. Grinding is carried out for 4 hours.

- L 2.5 g of piroxicam are mixed with 12.5 g of  $\beta$ -cyclodextrin for 15'. The powder is then poured into the grinding chamber of a centrifugal mill. Grinding is carried out for 4 hours.
  - $\underline{\text{M}}$  180.0 g of <u>nifedipine</u> are mixed with 900.0 g of  $\underline{\text{PVP}}$  for 15'. The powder is then poured into the grinding chamber of a high energy vibrational mill. Grinding is carried out for 4 hours.
- N 13.3 g of <u>UDCA</u> are mixed with 53.2 g of  $\beta$ -cyclodextrin for 15'. The powder is then poured into the grinding chamber of a low energy vibrational mill. Grinding is carried out for 4 hours.
- 15 <u>O 11.1 g of UDCA</u> are mixed with 55.5 g of  $\beta$ -cyclodextrin for 15'. The powder is then poured into the grinding chamber of a low energy vibrational mill. Grinding is carried out for 4 hours.

## LIST OF TABLES

- Table 1: Variation of  $\Delta H_f$  and of  $T_f$  of Nimesulide complex with different Carriers at 40°C and 75% of relative humidity (R.H.).
  - Table 2: Co-grinding test of Nimesulide with different Carriers.
- Table 3: Co-grinding test of Piroxicam with various Carriers (weight ratio piroxicam/carrier: 1/5).
  - Table 4: Co-grinding test of Nifedipine with various Carriers (weight ratio nifedipine/carrier: 1/5).
- Table 5: Co-grinding test of UDCA with various -30 Carriers.
  - Table 6: Co-grinding test of UDCA with various Carriers (weight ratio nimesulide/carrier: 1/4). Percentage release is given.

Table 7: Dissolution speed of Piroxicam co-ground with different Carriers (weight ratio nimesulide/carrier: 1/4). Percentage release is give.

### CHARACTERIZATION TESTS

- 5 Powders of co-ground products obtained by using NVP/VA carrier have been characterized by:
  - a). using differential scanning calorimetry
  - b). measuring dissolution speed

and compared with co-ground products obtained with commonly used carriers.

Operating conditions of the tests were the following: Differential scanning calorimetry

A differential scanning calorimeter Perkin Elmer, mod. Pyris 1, with nitrogen flow, and a heating speed of 10°C/min has been used.

## Measurement of dissolution speed

The method referred to in USP XXI, no. 2 has been used, using SOTAX apparatus, with thermostatation of dissolution means at 37°C, and a rotation speed of the blades of 100 rpm. The concentration of dissolved drug has been measured by means of a spectrophotometer (Perkin Elmer, mod. Lambda 25).

In the case of piroxicam as dissolution mean HCl 0.1N, pH 1.2 has been used; in the case of nimesulide phosphate buffer pH 7.5.

Table 1 contains data referring to melting enthalpy and temperature of co-ground products obtained from nimesulide with the carrier according to the present invention NVP/VA and with comparative carriers linear and reticulated PVP, under different conditions and conservation times: it can be observed that NVP/VA has a clearly higher ability in de-structuring nimesulide (lower melting enthalpies and temperatures) and in keeping said activation under the various conservation

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conditions.

Table 2 contains calorimetry data referring to coground products obtained from nimesulide with different carriers, with different weight ratios carrier/drug and different co-grinding times: in all cases co-ground products with NVP/VA are more destructured.

Table 3 contains calorimetry data referring to coground products obtained from piroxicam with different carriers (NVP/VA, PVP-CL, beta-cyclodextrin): also in this case NVP/VA carrier gives rise to lower melting enthalpies and temperatures.

In Table 4 said higher de-structuring ability of NVP/VA is shown for co-ground products with nifedipine, whereas Table 5 shows data referring to co-ground products obtained from ursodeoxycholic acid, which here again are more activated with NVP/VA carrier.

Unexpectedly again, higher dissolution speeds for coground products with NVP/VA with respect to commonly used carriers have been found, as shown in Table 6 for nimesulide and Table 7 for piroxicam. Said better dissolution properties are in line with the better properties of chemical-physical activation.

#### CLAIMS

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- 1. Method for preparing a composite product comprising a step in which an active substance undergoes cogrinding with a carrier comprising N-vinyl-2-pyrrolidone/vinyl acetate copolymer.
  - 2. Method according to claim 1, in which the carrier is N-vinyl-2-pyrrolidone/vinyl acetate.
  - 3. Method according to claim 1, in which the cogrinding step takes place in dry conditions.
- 4. Method according to claim 1, in which the active substance is chosen among non steroidal anti-inflammatory agents.
  - 5. Method according to claim 1, in which the active substance is chosen among anti-hypertensives.
- 6. Method according to claim 1, in which the active substance is chosen among hepato-biliary agents.
  - 7. Method according to claim 1, in which the active substance is chosen among substances that are scarcely soluble in water environment.
- 20 8. Method according to claim 7, in which the active substance is chosen among scarcely water soluble substances having a low dissolution speed.
  - 9. Method according to at least one of the preceding claims, in which the active substance is chosen among:
- anti-inflammatory agents, analgesics, relaxants, antimicrobic agents, antiseptics, acid pump inhibitors, H<sub>2</sub>
  antagonists, anti-emetics and anti-nausea, biliary
  acids, oral hypoglycemizers, diuretics, antihypertensives, sulfonamides, ace-inhibitors,
- hypolipemizers, anti-mycotic agents, antihistamines, hormones, quinolone derivates, antibacterial agents, beta-lactame and fluoroquinolone antibiotics, antiviral agents, anti-neoplastic agents, immuno-modulators and immuno-suppressors, anti-gout agents,

anesthetics, analgesics, antipyretics, 5HT<sub>1</sub> agonists, anti-Parkinson agents, anti-psychotic agents, tranquillizers, antidepressants, anti-parasitic agents, non-cortisone anti-allergic agents, anti-asthmatic agents, anti-glaucoma agents, inhibitors of carbonic anhydrase or beta-blockers.

- 10. Method according to claim 9, in which the active substance is chosen among: paracetamol, nifedipine, ibuprofen, sulindac, diclofenac, piroxicam, 10 alclofenac, ketorolac, indomethacine, naproxen, fenoprofen, flurbiprofen, ketoprofen, cimetidine, ranitidine, mesalazine, ursodeoxycholic mefenamic acid, sinvastatin, megestrol acetate, diazepam, cyclosporin, ubiquinone, lorazepam, 15 tolbutamide, ketanserine, furosemide, nicergoline, losartan, econazole, miconazole, taxol, progesterone, prednisolone, beclometasone, nalidixic finasteride, ciprofloxacine, ofloxacine, lomefloxacine, methotrexate, etoposide, daunorubicine, tamoxifen, allopurinol, clodronic acid, sumatriptan, carbamazepine, clorpromazine, clozapine, sulpiride, buspirone, fluoxetine, citalopram, caffeine, metronidazole, acetazolamide.
- 11. Method according to at least one of the preceding claims, in which the active substance and N-vinyl-2-pyrrolidone/vinyl acetate copolymer are present in a weight ratio between 1:200 and 10:1; preferably between 1:100 and 5:1.
  - 12. Composite product that can be obtained from a process according to at least one of the claims 1 to 11.
    - 13. Pharmaceutical composition comprising the composite product according to claim 12.
    - 14. Pharmaceutical composition according to claim 13,

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in which the pharmaceutical form is chosen among: tablet, capsule, pellet, syrup and solution.

- 15. Method for preparing the pharmaceutical composition according to claim 13 comprising a step in which the composite product according to claim 12 is mixed with excipients or pharmaceutically acceptable additives.
  - 16. Use of an active substance and of a carrier comprising N-vinyl-2-pyrrolidone/vinyl acetate for preparing a pharmaceutical formulation.

TABLE 1

			1/7					
	packet T and	$\mathbf{T}_{\mathbf{r}}(^{\circ}\mathbf{C})$	108.5	105.9	123.6	120.4	131.5	129.3
	60 days in packet at room T and RH	Δ <b>H</b> <sub>r</sub> (mJ/mg)	12.1	2.1	18.3	10.7	28.4	15.1
	s in et	$T_{r}(^{\circ}\mathbb{C})$	0	0	119.7	118.9	131.9	131.1
ns	40 days in packet	Δ <b>H</b> r (mJ/mg)	0	0	22.5	10.4	21.7	18.6
conditio	ı packet 25 days ı vial	$\mathrm{T_{f}}(^{\circ}\mathrm{C})$	0.	0	123.4	122.4	131.7	131.2
Conservation conditions	15 days in packet and then 25 days in open vial	AHr (mJ/mg)	. 0	0	19.0	16.8	22.5	20.6
Conse	packet 15 days closed	T <sub>r</sub> (°C)	107.8	0	117.6	119.2	132.8	132.0
	15 days in packet and then 25 days at 4°C in closed vial	ΔH <sub>r</sub> (mJ/mg)	4.4	0	18.9	18.2	25.9	19.9
	of sta-		107.0	105.0	121.9	119.2	130.0	128.2
	Beginning of sta- bility	ΔH <sub>r</sub> (mJ/mg) T <sub>r</sub> (°C)	20.4	10.2	22.8	22.0	21.0	21.0
	Nimesulide / Carrier Ratio		1/3	1/4	1/3	1/4	1/3	1/4
	Example		.1	2	Ą	В	C	α
	Carrier		NXDXXA	INVELVE.	arxa		DVP_CT	7.1.7

 $\Gamma ABLE$  2

		Nimesulide /			Activation	Activation time (hours)	(s		
Carrier	Example	Carrier Ratio	0		I		2		
		•	$\Delta \mathbf{H_f}  (\mathrm{mJ/mg})$	$T_{\mathbf{f}}(^{\circ}\mathbf{C})$	ΔH <sub>f</sub> (mJ/mg)	$T_{f}$ (°C)	AHr (mJ/mg)	$\Gamma_{f}(^{\circ}C)$	
NVP/VA	3	1/3	60.6	137.3	34.7	114.9	26.0	108.5	
	4	1/4	59.6	140.2	19.0	109.4	10.6	107.7	2/7
dΛd	Œ	1/3	95.8	149.8	28.5	136.0	21.0	135.4	
4	汪	1/4	75.9	149.4	15.0	133.0	15.1	130.5	
T)-qVq	· U	1/3	79.2	150.7	32.9	132.6	31.2	132.0	
	Ħ	1/4	77.2	150.5	24.6	130.6	23.7	129.6	<del></del>

TABLE 3

		:	,	<i>f</i>	Activation time (hours)	time (ho	ours)			
Example	)	0	ŗ.	ì	7	_	33			
<del>-</del>	Δ <b>H</b> <sub>r</sub> (mJ/mg)		$\Gamma_{\Gamma}(^{\circ}C)$ $\Delta H_{\Gamma}$ $(mJ/mg)$		$T_{\rm f}(^{\circ}{\rm C})$ $\Delta H_{\rm f}$ $({\rm mJ/mg})$	$T_f(^{\circ}C)$	$T_{\rm f}$ (°C) $\Lambda H_{\rm f}$ (mJ/mg)	$T_{f}(^{\circ}C)$	Δ <b>H</b> <sub>f</sub> (mJ/mg)	T <sub>f</sub> (°C)
	107.7	204.7	-	t	-		-	,	ľ	ı
	-		45.8	147.1	45.8	140.0	39.2	144.4	29.0	142.1
Ι	1	1	63.2	200.0	47.5	199.5	35.5	201.3	34.3	199.4
Т	1	•	62.3	177.9	64.9	177.5	9.99	177.8	54.5	176.9

3/7

TABLE 4

			<del></del>	4/	<del>,</del>
	#	T <sub>f</sub> (°C)	1	128.3	151.7
	•	Δ <b>H</b> <sub>f</sub> (mJ/mg)		15.8	33.7
		$T_{\mathbf{f}}(^{\circ}\mathbb{C})$	t	124.1 15.8	151.7
ours)		$T_f(^{\circ}C)$ $AH_f \over (mJ/mg)$ $T_f(^{\circ}C)$ $AH_f \over (mJ/mg)$ $T_f(^{\circ}C)$ $AH_f \over (mJ/mg)$ $T_f(^{\circ}C)$ $AH_f \over (mJ/mg)$	1	17.6	34.7
time (h	<b>(</b>	$T_{\rm f}({}^{\circ}{}_{\rm C})$	1	125.4	152.2
Activation time (hours)	•	Δ <b>H</b> <sub>f</sub> (mJ/mg)		17.8	41.2
A	[.	$^{\circ}$ $\mathbf{T}_{\mathbf{f}}$ (°C)	ŧ	128.9	153.0
,	Ţ	$\Delta \mathbf{H_f}$ (mJ/mg)	1	31.5	45.6
		$T_{\mathbf{f}}(^{\circ}\mathbf{C})$	174.3	1	,
	)	Δ <b>H</b> <sub>f</sub> (mJ/mg)	105.5	1	t
	Example		1	9	M
	Carrier		ı	NVP/VA	PVP

ABLE 5

						Time (hours)	ours)				
Carrier Carrier	,)	0.		I		2		3		4	
	AH <sub>r</sub> (mJ/mg		$T_{\mathbf{f}}({}^{\circ}\mathbb{C})$	$T_{f}(^{\circ}C) \left(\begin{array}{c c} \Delta H_{f} \\ (mJ/mg) \end{array}\right) T_{f}(^{\circ}C) \left(\begin{array}{c c} \Delta H_{f} \\ (mJ/mg) \end{array}\right) T_{f}(^{\circ}C) \left(\begin{array}{c c} \Delta H_{f} \\ (mJ/mg) \end{array}\right) T_{f}(^{\circ}C) \left(\begin{array}{c c} \Delta H_{f} \\ (mJ/mg) \end{array}\right)$	$T_{\mathbf{f}}(^{\circ}\mathbb{C})$	Δ <b>H</b> <sub>f</sub> (mJ/mg)	$T_{f}(^{\circ}\mathbb{C})$	Δ <b>H</b> <sub>f</sub> (mJ/mg)	$T_{f}(^{\circ}C)$	Δ <b>H</b> <sub>f</sub> (mJ/mg)	T <sub>f</sub> (°C)
1/4 20.3	20.3	İ	206.2	22.2	143.1	20.5	141.6	141.6 14.9	137.9	10.4	137.7
1/5 24.2	24.2		206.2.	20.7	140:1	16.3	139.1	9.2	138.2	5.8	135.5
N 1/4 81.1	81.1		207.5	40.2	202.5	5.7	201.5	4.2	203.5	7.9	204.8
1/5 84.1	84.1		207.5	37.4	203.0	13.3	205.8	5.7	207.3	5.9	201.1

 $\Gamma ABLE$  (

зыны	Hvamp		Time (minutes)	
	ordinary.	5	. 01	SI
NVP/VA	2	93.2 %	97.1%	%9.86
PVP	g	79.7 %	93.1 %	. %9:56
PVP-CL	α	40.6 %	65.2 %	75.8 %

ABLE 7

300			T.	Time (minutes)	tes)	
Callica	Evampre	2	4	9	8	. 0I
NVP/VA	5	49.3 %	90.4%	95.2 %	97.6 %	98.7 %
PVP-CL	I	45.4 %	63.2 %	75.5%	84.4 %	% 5:06

Internati lication No.

PCT/IT 03/00401 A. CLASSIFICATION OF SUBJECT MATTER
IPC 7 A61K9/14 According to International Patent Classification (IPC) or to both national classification and IPC **B. FIELDS SEARCHED** Minimum documentation searched (classification system followed by classification symbols) A61K Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched Electronic data base consulted during the international search (name of data base and, where practical, search terms used) EPO-Internal, WPI Data, PAJ, BIOSIS, EMBASE, MEDLINE C. DOCUMENTS CONSIDERED TO BE RELEVANT Citation of document, with indication, where appropriate, of the relevant passages Relevant to claim No. DE 199 29 361 A (BASF AG) 1-4,7-164 January 2001 (2001-01-04) column 1, lines 6,7 column 2, lines 38-54 column 3, lines 23-33 column 4, lines 25-31 X WO 01/68100 A (BREITENBACH JOERG ; 1-3,7-16MAEGERLEIN MARKUS (DE); HANTKE THOMAS (DE); ROSENB) 20 September 2001 (2001-09-20) page 2, lines 31-40 page 3, lines 28-30 page 5, lines 5-12,36-42 examples 1-7 Further documents are listed in the continuation of box C. Patent family members are listed in annex. . Special categories of cited documents: \*T\* later document published after the International filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the inventor. \*A\* document defining the general state of the art which is not considered to be of particular relevance invention \*E\* earlier document but published on or after the International "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone \*L\* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art. document referring to an oral disclosure, use, exhibition or other means \*P\* document published prior to the international filing date but later than the priority date claimed "&" document member of the same patent family Date of the actual completion of the International search Date of mailing of the international search report 11 March 2004 19/03/2004 Name and mailing address of the ISA Authorized officer European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,

Villa Riva, A

Fax: (+31-70) 340-3016

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